Lehman College  
City University of New York  
Department of Chemistry

General Chemistry Laboratory I      CHE-167  
Spring 2012

Instructor:

Office Hours:
Office:  
e-mail:

Course Description
CHE-167 - General Chemistry Laboratory I,  
One Laboratory of 3 hours and a Recitation of one hour are offered per week - Monday 12:30-1:30 Recitation 1:30-3:30 pm Laboratory  
4 hours / 2 credits  
Introduction to the practical aspects of chemical principles, with emphasis on quantitative measurements and analytical technique.

Corequisites: CHE 166

Place of course in degree program
This laboratory is a degree program requirement for Chemistry, Biochemistry and Biology program. This course is recommended to pre-medical, pre-veterinary, and pre-dental students.

Academic or Learning Objectives
Student Learning Outcomes: After completing this laboratory students should be able:  
a. to make precise and accurate measurements using physical and chemical equipment and instruments  
b. to use the measured data and theoretical concepts to solve problems.  
c. to use the mathematical and statistical analysis to assess the precision of the measurement.  
d. to understand the principles that govern chemical transformations, which include kinetics and equilibrium.

Required Readings
General Chemistry Laboratory Experiments, Staff of the Department of Chemistry, Lehman College Bronx, NY (Handouts)

Course Requirements and Grading
Each instructor will provide an addendum to the syllabus where the grading scheme will be detailed.

Attendance Policy
The attendance to the laboratory is compulsory. A student cannot miss more than TWO laboratories a. For the case of missing more than two laboratories the student will not
receive a passing grade. **No make-up laboratories will be given. This is in accordance with the chemistry department’s policy.**

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**Accommodating Disabilities**
Lehman College is committed to providing access to all programs and curricula to all students. Students with disabilities who may need classroom accommodations are encouraged to register with the Office of Student Disability Services. For more information, please contact the Office of Student Disability Services, Shuster Hall, Room 238, phone number, 718-960-8441.

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**The Science Learning Center (SLC)**
The Science Learning Center (SLC) is the tutoring center on campus. The SLC provides drop-in tutoring for natural and computer science courses. To obtain more information about the SLC, please visit their website at [http://www.lehman.edu/issp](http://www.lehman.edu/issp), or please call the ACE at 718-960-8175.

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**Classroom Policy:**

**Food policy:** FOOD and DRINKS are **STRICTLY PROHIBITED** in the chemistry laboratory.

**Cell Phone Policy:** Cell phones are disruptive, even in vibrate mode. Make sure your cell phones are in silent mode before class starts. Text-messaging during class is also highly disruptive (besides absolutely rude) and is forbidden. If a cell phone rings during class, lecture will be stopped, until the student will shut down the device and the following penalties are applicable:

- **5 pts penalty** if your cell phone rings while I am in class;
- **10 pts penalty** if you continue the disturbance (e.g., by letting it ring again);
- **15 pts penalty** for 1st ring on 2nd occasion;

**Electronic devices Policy:** No electronic devices can be used or kept accessible during laboratory experiments or during the examinations; this includes, but is not limited to i-Phones, cell-phones, beepers, iPods, MP3 players, tape-recorders, PDAs, **bluetooth** and other computing or music devices. Only **basic** calculators will be allowed. The **Graphic Calculators are not acceptable**.

**Required Equipment** *(to be provided by the student)*: padlock; detergent; paper towels; matches

**SAFETY GLASSES MUST BE WORN AT ALL TIMES IN LABORATORY!** The students without **SAFETY GLASSES will be not allowed to work in the laboratory**

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Academic Integrity

While honest scholarship is time-consuming and often requires hard work, it is also the primary process by which students learn to think for themselves. Faculty members must teach respect for methods of inquiry within the various disciplines and make assignments that will encourage honest scholarship; students in turn must uphold a standard of honesty within the College, thereby affirming the value and integrity of their Lehman degree. The following definitions and procedures govern cases involving undergraduate student work.

The most common forms of academic dishonesty are cheating and plagiarism. Cheating is the use or attempt to use unauthorized material, information, notes, study aids, devices, or communication during an academic exercise (for example, using unauthorized books, papers, or notes during an examination; or procuring, distributing, or using unauthorized copies of examinations). Plagiarism means the failure to give credit for the source of another's words or ideas, including but not limited to books, articles, interviews, and multimedia and electronic sites, or—as in the use of borrowed or purchased papers—passing off another person's work as one's own. (Section 213-b of the New York State Education Law prohibits the sale of term papers, essays, and research reports to students enrolled in a college.) Common forms of cheating and plagiarism are highlighted in this Bulletin.

Academic dishonesty is a serious violation of the accepted values of the College. When questions of a breach of academic integrity arise, instructors will inform the students of their suspicions and provide the student with a Faculty Report Form for Incidents of Suspected Academic Dishonesty. The instructor must remember that a student's failure to respond to charges of academic dishonesty is not in and of itself an indication of guilt. The report will include an explanation of the incident, the instructor's intended academic sanction, and an indication whether or not the instructor is recommending that the College undertake disciplinary proceedings pursuant to Article 15 of the Board of Trustees Bylaws.

Academic sanctions may include but are not limited to the following:

1. a grade of F for the course.

Disciplinary procedures are governed by Article 15 of the Board of Trustees Bylaws. In the event the student is found guilty of academic dishonesty by a Faculty-Student Disciplinary Committee, penalties that may be imposed include but are not limited to: 1) suspension from the College or 2) expulsion from the College. Although the Office of the Vice President for Student Affairs will be guided by the recommendation of the instructor, it reserves the right to seek disciplinary sanctions under the disciplinary procedures.

Should the instructor become convinced that the suspicions are unfounded, no further action will be taken and the Faculty Report Form will be destroyed. If the suspicions are founded and if both the student and the instructor are willing, they may agree upon a resolution. Subsequently the instructor will present the completed Faculty Report Form, including the charges and resolution, to the department chair who must forward the appropriate copies of the form to the Office of Academic Standards and Evaluation, and the Office of the Vice President for Student Affairs. If no agreement is reached, the instructor must allow a student to complete all coursework until the following appeal process has been completed.

- The first step in the appeals process is for the instructor to file the Faculty Report Form with the chair. If the term is completed, the instructor may assign a grade that reflects the intended sanction but must also provide a final grade that does not include the intended sanction if the charges are not upheld.
- If the charges are for cheating, then the chair will submit the charges to the Office of the Vice President for Student Affairs. If the charges are for plagiarism, the chair will appoint a committee of three Lehman College faculty members, which will adjudicate the matter within three weeks by majority vote. If the chair is the instructor in question, the senior member of the department
Personnel and Budget Committee will act for the chair. The committee will provide written notification of its decision to the chair, who will forward this recommendation and the Faculty Report Form to the Office of the Vice President for Student Affairs.

- The Office of the Vice President for Student Affairs will review the recommendations of the instructor and the committee for possible disciplinary sanctions and provide a written notification of its decision to the department chair, the student, the instructor, and the Office of Academic Standards and Evaluation. Either the instructor or the student has the right, within three weeks of receipt of notification, to appeal the department decision in writing to the Committee on Admissions, Evaluation, and Academic Standards, which will act as adjudicator of last resort. Should any part of the three-week period fall outside the regular semester, the first three weeks of the next regular semester shall apply.

The Office of Academic Standards and Evaluation will keep all records of such proceedings on file until the student's graduation, at which time they will be destroyed.

As a result of a second upheld charge of academic dishonesty, disciplinary procedures will be pursued by the Office of the Vice President for Student Affairs as governed by the procedures under Article 15 of the Board of Trustees' Bylaws.

**The following definitions and examples are adapted from the CUNY Policy on Academic Integrity.**

**Cheating** is the unauthorized use or attempted use of material, information, notes, study aids, devices, or communication during an academic exercise. Examples of cheating include, but are not limited to the following:

- Copying from another student during an examination or allowing another student to copy your work.
- Unauthorized collaboration on a take-home assignment or examination.
- Using illegal notes during a closed-book examination.
- Taking an examination for another student, or asking or allowing another student to take an examination for you.
- Changing a graded exam and returning it for more credit.
- Submitting substantial portions of the same paper for more than one course without informing each instructor.
- Preparing answers or writing notes in a blue book (exam booklet) before an examination.
- Allowing others to research and write assigned papers or do assigned projects, including the use of commercial term paper services.
- Giving assistance to acts of academic misconduct or dishonesty.
- Fabricating data (all or in part).
- Submitting someone else's work as your own.
- Unauthorized use during an examination of any electronic devices, such as cell phones, palm pilots, computers, or other technologies to send or retrieve information.

**Plagiarism** is the act of presenting another person's ideas, research, or writings as your own. Examples of plagiarism include, but are not limited to the following:

- Copying another person's actual words without the use of quotation marks and citations.
- Presenting another person's ideas or theories in your own words without acknowledging the source.
- Using information that is not common knowledge without acknowledging the source.
- Failing to acknowledge collaborators on assignments.
- Purchasing or downloading term papers online.
- Paraphrasing or copying information from the Internet without citing the source.
- "Cutting and pasting" from various sources without proper attribution.
Course topics
The following topics will be covered:
Mass and Volume Relationships; Separation; Reactions of Copper; Determination of the Formula of a Compound Analysis of a Hydrate; Limiting Reagent; Titration; Conductivity; Redox reaction; Types of Bonding; UV-VIS spectroscopy.

Homework Exercise: Each laboratory contains a Pre-lab assignment. The Pre-Lab assignment should be return to the instructor before beginning of each laboratory.

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<td>2</td>
<td>1B</td>
<td>Mass and Volume Relationships</td>
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<td>Separation of the Components of a Mixture</td>
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<td>Reactions of Copper</td>
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<td>Determination of the Formula of a Compound</td>
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<td>Analysis of a Hydrate Part I</td>
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<td>Analysis of a Hydrate Part II</td>
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<td>Chemical Reactions and Ionic Equations</td>
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<td>Limiting Reagent</td>
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<td>Titration: Volumetric Analysis of Acid and Base</td>
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<td>Oxidation-Reduction Titrations</td>
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<td>Electrical Conductivity of the Solutions</td>
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<td>Determining the Concentration of a Solution: Beer’s Law</td>
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# Spring 2012

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